



# Course Specification

## (Bachelor)

**Course Title:** Computer Applications in Chemical Laboratories

**Course Code:** CHM 1153

**Program:** Bachelor of Science in Chemical Laboratories

**Department:** Chemistry

**College:** Science

**Institution:** Imam Mohammed Ibn Saud Islamic University

**Version:** 2024- 1

**Last Revision Date:** 15 September 2024



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## A. General information about the course:

### -1. Course Identification

1. Credit hours: 2 (0, 0, 4)

2 (0 Lectures, 0 Tutorials, 4 Lab)

2. Course type

A ☐ University ☐ College ☒ Department ☐ Track ☐ Others

B ☒ Required ☐ Elective

3. Level/year at which this course is offered: Level 2/ Second year

4. Course general Description:

The course covers all required knowledge and information for appropriate software that will be used in labs and courses. It will provide sufficient information and application to perform searches, use software associated with lab instruments, and write reports and data.

5. Pre-requirements for this course (if any):

General Chemistry 1 - CHM 1101

6. Co-requisites for this course (if any):

None

7. Course Main Objective(s):

At the end of this course the student will be able to:

- Analyze the various chemical databases
- Use the computer software such as Excel to present his data graphically and obtain constants
- Utilize the computational tools to solve the chemical problems
- Apply various associated software with spectroscopic lab instruments
- Apply various computational methods to solve chemical problems

### 2. Teaching mode (mark all that apply)

No	Mode of Instruction	Contact Hours	Percentage
1	Traditional classroom	60	100%
2	E-learning	0	0
3	Hybrid <ul style="list-style-type: none"> <li>Traditional classroom</li> <li>E-learning</li> </ul>	0	0
4	Distance learning	0	0



### 3. Contact Hours (based on the academic semester)

No	Activity	Contact Hours
1.	Lectures	0
2.	Laboratory/Studio	60
3.	Field	0
4.	Tutorial	0
5.	Others (specify)	0
Total		60

### B. Course Learning Outcomes (CLOs), Teaching Strategies and Assessment Methods

Code	Course Learning Outcomes	Code of PLOs aligned with program	Teaching Strategies	Assessment Methods
1.0	Knowledge and understanding			
1.1	To identify the importance of software chemistry	K1; K2; K4	<ul style="list-style-type: none"> <li>• Four hours weekly, containing laboratory activities and oral discussions.</li> <li>• A Private study, including work on writing a report.</li> </ul>	<b>Direct:</b> <ul style="list-style-type: none"> <li>• Laboratory Reports.</li> <li>• Oral Discussion marks</li> <li>• Participation.</li> </ul>
1.2	To list of applications of software chemistry	K1; K2; K4	<ul style="list-style-type: none"> <li>• Four hours weekly containing laboratory activities with group discussion. Think and discuss the Required Software chemistry</li> </ul>	<b>Direct:</b> <ul style="list-style-type: none"> <li>• Lab. Reports.</li> <li>• Oral Discussions.</li> </ul>
1.3	To recognize scientific journals and database.	K1; K2; K4	<ul style="list-style-type: none"> <li>• Four hours are weekly containing Laboratory activities with group discussion.</li> <li>• Think and write about the chemical equation by</li> </ul>	<b>Direct:</b> <ul style="list-style-type: none"> <li>• Laboratory Reports</li> <li>• Oral Discussions.</li> </ul>



Code	Course Learning Outcomes	Code of PLOs aligned with program	Teaching Strategies	Assessment Methods
			appropriate software	
2.0	<b>Skills</b>			
2.1	To illustrate a simple mechanism using appropriate software program	S1; S2	<ul style="list-style-type: none"> <li>Laboratory activities</li> <li>Think and write about the chemical equation by appropriate software</li> </ul>	<b>Direct:</b> <ul style="list-style-type: none"> <li>Questions in labs.</li> <li>Participation through Laboratories</li> <li>Oral Discussion,</li> </ul>
2.2	To compare between the software programs	S1; S2	Encourage students to communicate their logic chemical thinking, and to work and discuss cooperatively with their peers to develop individual skills.	<b>Direct:</b> <ul style="list-style-type: none"> <li>Questions in labs.</li> <li>Participation through Laboratories</li> <li>Oral discussion</li> </ul>
2.3	To use Computer, software in, Perform Calculations, and chemical drawing.	S4	Encourage the students to use the Chemicals Glass wares and Instruments with caring and safety	<b>Direct:</b> <ul style="list-style-type: none"> <li>Oral Discussion.</li> <li>Discussion marks Giving marks for participation in the lab.</li> </ul>
3.0	<b>Values, autonomy, and responsibility</b>			
3.1	To show self-confidence attitudes through single and team work practical sessions, presentations, and discussions.	V1, V2	<ul style="list-style-type: none"> <li>labs and Group discussion</li> <li>Have the ability to ask and answer questions as they arise</li> <li>Brain storming Exercises</li> </ul>	<b>Direct:</b> <ul style="list-style-type: none"> <li>Questions in labs.</li> <li>Participation through Laboratories</li> <li>Oral discussion.</li> </ul>





### C. Course Content

No	List of Topics	Contact Hours
1.	Introduction to computer: Introduction and Fundamentals of Computer, MS office and applications, basic commands, handling Cookies.	10
2.	Chemistry software's: Introduction to chemoffice, Chems sketch tools to draw chemical structures. Convert chemical name to structure and vice versa, Chemical data bases like CAS, Pubmed,	20
3	Microsoft Excel, Elementary structural features such as bond lengths, bond angles, dihedral angles etc., of molecules.	15
4	Bibliographic Databases using Endnote, Mendeley, Chemical Abstracts, Journals, Conferences, Reports, Patents, Softwares associated with spectroscopic instruments, origin, sigmaplot.	15
Total		60

### D. Students Assessment Activities

No	Assessment Activities *	Assessment timing (in week no)	Percentage of Total Assessment Score
1.	Midterm 1	6 <sup>th</sup> / 7 <sup>th</sup> week	20 %
2.	Midterm 2	11 <sup>th</sup> / 12 <sup>th</sup> week	20 %
3.	Quizzes, Home Works, class participation, and mini projects	During the semester	20 %
5.	Final Exam	16 <sup>th</sup> week	40 %
6.	Total	All weeks	100 %

\*Assessment Activities (i.e., Written test, oral test, oral presentation, group project, essay, etc.).

### E. Learning Resources and Facilities

#### 1. References and Learning Resources

Essential References	Peter C. Jurs, , <b>Computer Software Applications in Chemistry</b> , 2 <sup>nd</sup> Edition. ISBN: 978-0-471-10587-9
Supportive References	Blackboard





Electronic Materials

Other Learning Materials

## 2. Required Facilities and equipment

Items	Resources
<b>facilities</b> (Classrooms, laboratories, exhibition rooms, simulation rooms, etc.)	<b>Each Laboratory should be equipped with maximum 20 seats</b>
<b>Technology equipment</b> (projector, smart board, software)	<b>The rooms are equipped with data show, Smart Board, WI-FI access.</b>
<b>Other equipment</b> (depending on the nature of the specialty)	<b>Software: Chem Draw, office (Words, Excel), ACD labs Computers</b>

## F. Assessment of Course Quality

Assessment Areas/Issues	Assessor	Assessment Methods
Effectiveness of teaching	<b>Students</b>	<b>Direct:</b> Questionnaire.
	<b>Course Responsible</b>	<b>Direct:</b> Course e-Portfolio.
	<b>Peer Reviewer</b>	<b>Indirect:</b> Second examiner checklist-Course report.
Effectiveness of Students assessment	<b>Program Leaders</b>	<b>Direct:</b> Questionnaire.
	<b>Students</b>	<b>Indirect:</b> External assessor report.
Quality of learning resources	<b>Faculty ( Academic Advisory)</b>	<b>Direct:</b> Course e-Portfolio.
	<b>Program Leaders</b>	<b>Indirect:</b> Course report.
		<b>Indirect:</b> Second examiner checklist-Course report.
		<b>Direct:</b> course Entrance/Exit.
		<b>Indirect:</b> Observations - Accreditation review.
		<b>Direct:</b> Course e-Portfolio.
		<b>Indirect:</b> Course evaluation survey- Observations- Syllabus review- Accreditation review.
The extent to which CLOs have been achieved		
Other		

**Assessors** (Students, Faculty, Program Leaders, Peer Reviewer, Others (specify))

**Assessment Methods** (Direct, Indirect)





### G. Specification Approval

COUNCIL /COMMITTEE	COUNCIL OF DEPARTMENT OF CHEMISTRY
REFERENCE NO.	3 (NO. 1/3)
DATE	5/3/1446- 8/09/2024

